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Research

Design and invitro characterization of alfuzosin HCL tablets for sublingual drug delivery system

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	Abstract
Published on: 25 Sept 2024	<p>Alfuzosin HCl (ALF) that is commonly used to treat benign prostatic hyperplasia (BPH), on the market. On the other hand, ALF-containing orally disintegrating tablets (ODTs) are not yet available for purchase. To make the selected medicinal molecule more soluble and to speed up its dissolving rate, this study developed sublingual tablets of ALF employing super disintegrating agents such as sodium starch glycollate, locust bean gum, and cross carmellose sodium. All of the formulations were prepared using the direct compression method with a 6-millimeter punch on an 8-station rotary tablet punching machine. All things considered, the mixtures displayed good flow properties including bulk density, tapped density, and angle of repose. All of the quality control evaluation parameters were met by the produced tablets, and they demonstrated good post compression properties according to the I.P limits. Optimal performance was achieved by the AF1 formulation, which exhibited the highest percentage of drug release (99.78%) in under 6 minutes. A super disintegrates containing ten milligrams of SSG is included in the AF1 formulation. According to the results of the permeability tests, ALF is an easily absorbed substance. Based on these results, the ODT method for ALF seems to be a good way to get patients to comply with treatment and get things rolling quickly.</p>
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	<p>Keywords: Alfuzosin HCl, Sodium starch glycollate, Locust bean gum and Cross carmellose sodium.</p>

INTRODUCTION

The need for a quick pharmacological effect is what gave rise to the sublingual route of drug delivery. Patients of all ages can experience dysphasia, or trouble swallowing; however, those who are cognitively impaired,

resistant to medication, sick to their stomach, or on restricted liquid diets may find it particularly challenging to swallow these dosage forms. The region under the tongue has the highest porosity when the entire buccal cavity is consideredⁱ. There is usually a faster onset of effect when administered sublingually rather than orally because the absorption through the sublingual blood vessels usually happens before the hepatic first-pass metabolic processes. Sublingual tablets are usually small, flat, and lightly compressed so that they retain their softness. The tablet needs to dissolve in just a small amount of saliva for the active pharmaceutical ingredient (API) to be absorbed quickly. Developing systemic drug administration via the sublingual method was driven by the need for rapid pharmacological activity. With a medication absorption rate three to ten times greater than that of the oral route, the sublingual technique ranks second only to intradermal injection. Sublingual administration is the best option for drugs having a short half-lifeⁱⁱ. There were some sublingual drug delivery formulations developed earlier such as Losartan potassiumⁱⁱⁱ, carvedilol^{iv}, rizatriptan benzoate^v, felodipine^{vi}, glimepiride^{vii}, perindopril^{viii}, ondansetron^{ix}.

Locust bean is a natural polymer. It is utilized in the development of several drug delivery systems such as tablets, capsules, beads, microspheres, gel and polymeric films. Hence we used locust bean gum in this research. Alfuzosin is a white to off-white crystalline powder. Soluble in water, sparingly soluble in alcohol, and practically insoluble in dichloromethane. ALF is used to treat the signs and symptoms of benign prostatic hyperplasia (BPH). Stored in closed container at room temperature, avoid direct sunlight^x. In this study, we tried to formulate sublingual tablets of ALF HCl. The current work utilized super disintegrating agents including sodium starch glycolate, locust bean gum, and cross carmellose sodium to enhance the solubility and rate of dissolution of a specific pharmaceutical molecule.

MATERIALS & METHODS

Chemicals

Terbutaline sulfate was obtained as a gift sample from Natco Pharmaceuticals, Hyderabad, India. All other chemicals such as sodium starch glycolate, cross carmellose sodium, magnesium stearate, locust bean gum, microcrystalline cellulose, talc, sodium saccharine and mannitol were purchased from SD Fine Chemicals, Mumbai. All the reagents and chemicals used were of analytical grade.

METHODOLOGY

Determination of absorption maxima

A double beam UV/VIS spectrophotometer is used to obtain the UV spectra of a phosphate buffer with a pH of 6.8 buffer solution containing 10 µg/ml of the ALF has been made. The 200-400 range was used to scan the solution^{xi}.

Drug- excipient compatibility studies by FT-IR^{xii}

The compatibility of the pure medicine and excipients might be ascertained by the use of FTIR spectra recorded on a Shimadzu FTIR spectrometer. The solid powder sample was ground with a mortar and KBr at a rate one hundred times the original amount in order to produce potassium bromide pellets by means of a KBr press. Afterwards, the powder that had been finely ground was placed into a die made of stainless steel and compressed between anvils of polished steel at a pressure of around 8t/in². The spectra encompassed a wave number range of 4000 to 400 cm⁻¹.

Formulation of Sublingual tablets of ALF^{xiii}

The fast-dissolving sublingual tablets containing 10 mg of ALF were manufactured via direct compression, meaning that no granules were required for their production. According to the data shown in Table 1. Dry mixing, lubrication, compression, and sieving are the four steps that make up the direct compression method of making tablets. Each tablet contains several medical ingredients and 10 milligrams of ALF. It was found that the entire weight of the tablet was 150 mg^{xiv}.

Table 1: Composition of various tablet formulations

Ingredients	AF1	AF2	AF3	AF4	AF5	AF6	AF7	AF8	AF9
ALF (mg)	10	10	10	10	10	10	10	10	10
CCS (mg)	10	15	20	-	-	-	-	-	-
SSG (mg)	-	-	-	10	15	20	-	-	-
Locust bean gum (mg)	-	-	-	-	-	-	10	15	20
Mg Stearate(mg)	5	5	5	5	5	5	5	5	5

Talc(mg)	5	5	5	5	5	5	5	5	5
MCC (mg)	20	20	20	20	20	20	20	20	20
Mannitol	45	45	45	45	45	45	45	45	45
Sodium saccharin	10	10	10	10	10	10	10	10	10
Total wt. (mg)	150	150	150	150	150	150	150	150	150

Evaluation of Tablets^{xv, xvi}

The formulated tablets were evaluated for the physicochemical parameters such as thickness, hardness, friability, weight variation test and uniformity of drug content.

In-vitro dissolution studies

The in-vitro release investigations were conducted using a Lab India DS-800, a modified USP XXIII dissolving test device. In each test, a dissolving fluid of 500 milliliters of pH 6.8 phosphate buffer was utilized, along with a speed of 50 rpm and a temperature of 37°C. To find out if ALF was present, we tested the absorbance of 5 ml samples of the dissolving liquid at 254 nm every two minutes. At regular intervals, five milliliters of the test medium were withdrawn and replaced with the same volume of pH 6.8 phosphate buffer for each test^{xvii}.

Release kinetics

Plotting the data from the in vitro dissolution investigation in several kinetics models allowed researchers to examine the sublingual tablets' in-vitro release kinetics^{xviii}.

Stability studies of the tablet

In this study, the selected formulation (AF1) was subjected to 12-week stability tests at 25°C/60%RH and 40°C/75%RH^{xix}. We kept the selective packed formulations at 25°C with 60% relative humidity and 40°C with 75% relative humidity for 12 weeks. The medication content and the formulation's physical appearance were evaluated at set intervals. The formulations underwent additional scanning to search for possible alterations in the spectrum. Furthermore, studies on solubility in vitro were carried out^{xx}.

RESULTS AND DISCUSSION

Standard Calibration curve of TBS

Because of its high reproducibility, the UV spectrophotometric method was used to measure ALF at λ_{\max} 254 nm in a pH 6.8 phosphate buffer in this investigation. A correlation coefficient closer to 1 was observed for the standard curve in the concentration range of 0.1-0.6 $\mu\text{g/ml}$. A regression equation was generated with an R^2 value of 0.9983, which is $y = 1.2809x + 0.0126$.

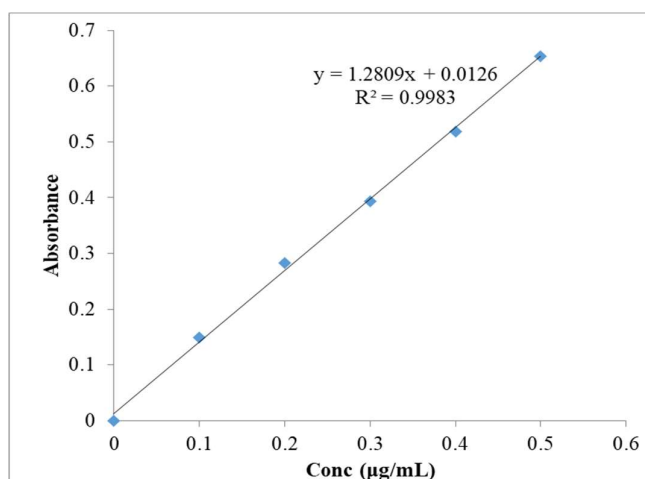
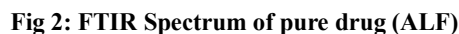


Fig 1: Standard graph of ALF in pH 6.8 Phosphate buffer

Evaluation Parameters for Sublingual Tablets of ALF

FTIR

The drug, polymer, and physical mixture were all graphed in FTIR; the physical combination shouldn't have any displacement of the functional peaks. It is very evident that the medicine and excipient did not interact.



Pre-compression parameters

Table 2: Results of derived and flow properties of ALF physical mixture277

AF3	0.61	0.71	19.72	1.217	24.22
AF4	0.59	0.73	26.16	1.335	23.11
AF5	0.61	0.69	20.04	1.222	21.95
AF6	0.56	0.67	24.29	1.29	23.17
AF7	0.59	0.67	18.93	1.205	21.4
AF8	0.62	0.75	22.81	1.294	19.11
AF9	0.58	0.68	22.13	1.254	21.97

All values are expressed as mean \pm S.D, n = 3

Post compression Parameters

Hardness test

The table displays the results of a Monsanto toughness tester that was used to analyze the three tablets from each batch for hardness. Results showed that the tablet hardness, which varied from 2.85 to 3.43 kg/cm², was within the acceptable range for IP requirements. This ensures that every batch will perform admirably^{xxiii}.

Thickness

The three tablets from all of the batches were tested for thickness using a Vernier Caliper, according to the data in Table 3. The findings indicate that the thickness of the tablet varies between 2.84 and 3.29 mm^{xxiv}.

Weight variation test

Table 15 displays the results of a measurement of weight variation test that was conducted on each batch of tablets. The allowed limit is $\pm 10\%$, because the average weight of the tablet is between 94.65 and 101.25 mg. The results of the tests showed that the tablet weights were within the acceptable range for the pharmacopoeia^{xxv}.

Table 3: Results of average weight, hardness, thickness, friability of all formulations ALF sublingual tablet

F. Code	Weight variation (mg)	Hardness (kg/cm ²)	Thickness (mm)	Friability (%)	Assay (%)
AF1	97.42	2.91	3.02	0.45	99.67
AF2	100.21	2.85	3.29	0.51	98.73
AF3	94.65	3.18	3.21	0.69	97.85
AF4	95.62	3.05	2.95	0.78	96.56
AF5	98.53	2.87	2.84	0.48	97.23
AF6	99.31	3.1	3.03	0.69	95.67
AF7	101.25	3.25	3.17	0.89	98.57
AF8	97.54	3.43	3.26	0.92	97.63
AF9	95.22	3.15	3.05	0.78	96.55

All values are expressed as mean \pm S.D, n = 3

Friability

The statistics regarding the % friability of each batch's tablets is displayed in Table. A satisfactory mechanical toughness of tablets was shown by an average friability ranging from 0.45 to 0.92 percent, which was below the 1% official criterion of IP^{xxvi}.

Assay

Research using assay methods was performed on the pre-prepared mixes. All of the formulations showed drug content values ranging from 95.67% to 99.67%, according to the lab tests^{xxvii}.

In vitro disintegration time

The in vitro disintegration time of each batch's tablets was measured and the findings were displayed in a table. Results showed that the manufactured tablets had disintegration durations between 44.05 and 77.71 seconds^{xxviii}.

Table 4: Results of water absorption ratio, drug content, wetting time, disintegration of all formulations

F. Code	Water absorption ratio (%)	Drug Content (%)	Wetting time (Sec)	Disintegration time (Sec)
AF1	25.31	99.43	7.48	44.05
AF2	27.72	97.33	6.74	51.06

AF3	29.05	99.99	8.65	58.55
AF4	28.64	96.61	9.03	62.23
AF5	34.24	97.33	11.14	67.18
AF6	30.2	94.01	10.06	66.21
AF7	37.54	95.43	13.15	72.15
AF8	38.62	97.57	11.32	77.71
AF9	38.99	96.06	9.88	77

All values are expressed as mean \pm S.D, n = 3

Drug content: The drug concentration was found to be between 94.01% and 99.99%. The results were presented in a table, and the results show that they may be reproduced with little variation from batch to batch^{xxix}.

Wetting time: Table contains a tabulation of the wetting time values. The range of the wetting time was 7.48 to 13.15 seconds.

In vitro Dissolution studies

In vitro dissolution tests were carried out using the paddle method with a USP dissolving device and 500 milliliters of pH 6.8 phosphate buffer. Almost eight minutes passed during the dissolving experiments. The fastest drug release rate (99.78%) within 6 minutes was observed in the AF1 formulations with 10 mg of super disintegrate cross carmellose sodium, as shown clearly in the tabular column. Consequently, it was found that sublingual tablets can be made using the super disintegrates approach. The AF1 formulation was considered to be the best.

Application of Release Rate Kinetics to Dissolution Data

Multiple models were employed to examine the drug release kinetics. In order to investigate the mechanism of the dosage form's drug release rate kinetics, the acquired data were fitted into the Higuchi, Korsmeyer-Peppas, zero-order, and first-order release models.

Stability Studies

For a duration of three months, an optimized formulation, AF-1, was the subject of stability investigations. Although there has been a small rise in assay, hardness, and friability, the overall look has not changed. At 25°C \pm 2°C/60% \pm 5% RH and 40°C \pm 2°C/75% \pm 5% RH, the dissolving profile of the optimum batch was calculated. The results showed that at each temperature, the percentage of drug released reduced marginally.

CONCLUSION

In this study, we tried to formulate ALF HCl tablets to be taken under the tongue. The current work utilized super disintegrating agents including sodium starch glycollate, locust bean gum, and cross carmellose sodium to enhance the solubility and rate of dissolution of a specific pharmaceutical molecule. Every single one of the recipes was made by compression using a rotary tablet punching machine that had eight stations and a 6mm punch. The bulk density, tapped density, and angle of repose were all well-represented by the combined formulations. The manufactured tablets fulfilled all I.P. limits for every quality control evaluation metric and showed good post-compression characteristics. Among all the formulations, AF1 showed the greatest percentage of drug release 99.78 percent in 6 minutes and is thus considered the best.

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