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## Analytical method development for the estimation of everolimus by rp-hplc

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## **ABSTRACT**

In reversed phase systems the strong attractive forces between water molecules arising from the 3-dimentional inter molecular hydrogen bonded network, from a structure of water that must be distorted or disrupted when a solute is dissolved. Only higher polar or ionic solutes can interact with the water structure. Non-polar solutes are squeezed out of the mobile phase and are relatively insoluble in it but with the hydrocarbon moieties of the stationary phase.

A sensitive & selective RP-HPLC method has been developed & validated for the analysis of Everolimus API. Further the proposed RP-HPLC method has excellent sensitivity, precision and reproducibility.

Keywords: RP-HPLC, Everolimus, Method development, Method validation

## INTRODUCTION

Reversed Phase Chromatography Since 1960's chromatographers started modifying the polar nature of silanol group by chemically reacting silica with organic silanes. The objective was to make less polar or non polar so that polar solvents can be used to separate water-soluble polar compounds. Since the ionic nature of the chemically modified silica is now reversed i.e. it is non-polar or the nature of the phase is reversed. The chromatographic separation carried out with such silica is referred to as reversed- phase chromatography.

As a general rule the retention increases with increasing contact area between sample molecule and stationary phase i.e. with increasing number of water molecules, which are released during the adsorption of a compound. Branched chain compounds are eluted more rapidly than their corresponding normal isomers.

In reversed phase systems the strong attractive forces between water molecules arising from the 3-dimentional inter molecular hydrogen bonded network, from a structure of water that must be distorted or disrupted when a solute is dissolved. Only higher polar or ionic solutes can interact with the water structure. Non-polar solutes are squeezed out of the mobile phase and are relatively insoluble in it but with the hydrocarbon moieties of the stationary phase.

# Adsorption Chromatography or Normal Phase Chromatography<sup>[2,3]</sup>

In normal phase chromatography, the stationary phase is a polar adsorbent and the mobile phase is generally a mixture of non-aqueous solvents. Chromatographic methods can be classified most practically according to the stationary and mobile phases, as shown in the table-1.

## Classification of Chromatographic methods

Stationary phase	Mobile phase	Method		
Solid	Liquid	Adsorption column, thin-layer, ion exchange, High		
		performance liquid chromatography.		
	Liquid	Partition, column, thin-layer, HPLC, paper		
Liquid		chromatography.		
	Gas	Gas – Liquid Chromatography.		

The importance of Chromatography is increasing rapidly in pharmaceutical analysis. The exact differentiation, selective identification and quantitative determination of structurally closely related compounds are possible with chromatography. Another important field of application of chromatographic methods is the purity testing of final products and intermediates (detection of decomposition

products and by-products). As a consequence of the above points, chromatographic methods are occupying an ever-expanding position in the latest editions of the pharmacopoeias and other testing standards.

The various components of a HPLC system are here with described:

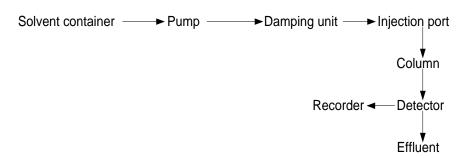
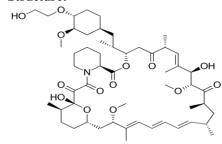


Fig 1: Various Components of HPLC

## **DRUG PROFILE**

Name: EVEROLIMUS

## **Structure:**



**Molecular Weight:** 958.24 **Chemical Formula:** C<sub>53</sub>H<sub>83</sub>NO<sub>14</sub>

**Indication**: Everolimus is indicated for the treatment of postmenopausal women with advanced hormone receptor-positive, HER2-negative breast cancer (advanced HR+ BC) in combination with exemestane, neuroendocrine tumors of pancreatic origin (PNET) with unrespectable, locally advanced or metastatic disease. renal cell carcinoma (RCC). renal angiomyolipoma and tuberous sclerosis complex (TSC), tuberous sclerosis complex (TSC) for the treatment of subependymal giant cell astrocytoma (SEGA) that requires therapeutic intervention but cannot be curatively resected.

## **METHOD DEVELOPMENT**

## **Materials and Instruments:**

## **Equipments:**

A suitable HPLC having isocratic system equipped with manual injector with UV detector. Analytical Balance, capable of measuring the 0.01mg, Sonicator, Usual laboratory glass ware of class-A

## **Chemicals and Reagents:**

S.N.	Name	Specifications		Manufacturer/Supplier
		Purity	Grade	_
1.	HPLC grade water			Sd fine-Chem ltd; Mumbai
2.	Methanol	99.9%	A.R.	Loba Chem; Mumbai.
3.	Dipotassium hydrogen orthophosphate	96%	L.R.	Sd fine-Chem ltd; Mumbai
4.	Acetonitrile	99.9%	HPLC	Loba Chem; Mumbai.
5.	Potassium dihydrogen orthophosphate	99.9	L.R.	Sd fine-Chem ltd; Mumbai

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6.	Ortho phosphoric acid	99.9	L.R.	Sd fine-Chem ltd; Mumbai
7.	3% Hydrogen Peroxide	99.9%	A.R.	Vijaya Enterprises

## **Solubility**

REAGENTS	SOLUBILITY
DMSO	Soluble
Ethanol	Soluble
Dimethyl Formamide (DMF)	Soluble
Aqueous Buffers	Sparingly Soluble
Acetonitrile	Insoluble
Water	Insoluble

**HPLC Instrumentation & Conditions:** The HPLC system employed was **HITACHI L2130** with D Elite 2000 Software with Isocratic with UV-Visible Detector (L-2400).

**Mobile phase preparation:** The mobile phase used in this analysis consists of a mixture of Methanol and Acetronitrile in a ratio of 30:70.

Sample & Standard Preparation of solutions: Working concentration should be around 10 µg/ml.Accurately

weighed around 10mg of everolimus working standard, taken into a 100 ml volumetric flask, then dissolved and diluted to volume with the mobile phase to obtain a solution having a known concentration of about 1000 mcg/ml. Further dilutions has been made to get the final concentration of 10  $\mu g/ml$  Diluted quantitatively an accurately measured volume of label claim solution with diluents to obtain a solution containing about a linear range.

**Table 1: Trials For Method Development and Optimized Condition** 

Column Used	Mobile Phase	Flow Rate	Wave length	Observation	Result
C18 Develosil ODS HG-5 RP 150mm x	Methanol :	1.0 ml/min	272nm	Low	Method
4.6mm 5µm particle size	Water = 70:30			response	rejected
C18 Develosil ODS HG-5 RP 150mm x	ACN only	0.5 ml/min	272nm	Very low	Method
4.6mm 5µm particle size				response	rejected
C18 Develosil ODS HG-5 RP 150mm x	ACN: water	1.0 ml/min	272nm	Peak broken	Method
4.6mm 5µm particle size	= 70:30				rejected
C18 Develosil ODS HG-5 RP 150mm x	ACN: acetate	1.0	272nm	Broad Peak	Method
4.6mm 5µm particle size	buffer $= 70:30$	ml/ min			rejected
C18 Develosil ODS HG-5 RP 150mm x	Methanol:	1.0 ml/min	272nm	Good sharp	Method
4.6mm 5µm particle size	Acetonitrile =			peak	accepted
	30:70				

## **Optimized Chromatographic Conditions**

Column : C<sub>18</sub> Develosil ODS HG-5 RP 150mm x 4.6mm 5µm particle size

Mobile Phase : Methanol : Acetonitrile (30:70)

Flow Rate : 1.0ml/minute
Wave length : 272 nm
Injection volume : 10 µl
Run time : 10 minutes
Column temperature : Ambient
Sampler cooler : Ambient

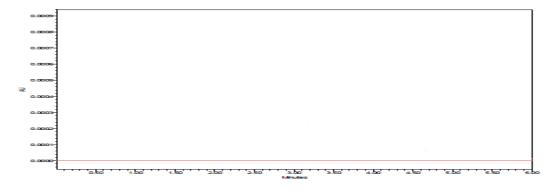


Fig 2: Chromatogram for Blank

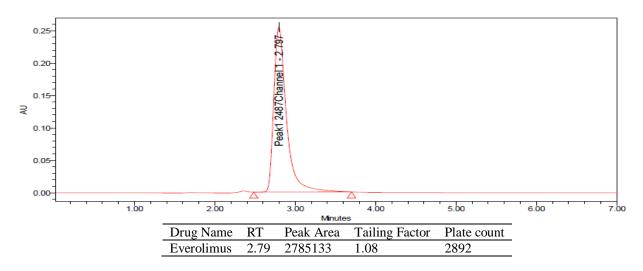


Fig 3: Chromatogram for Optimised Condition (Everolimus) (Rt 2.79)

## **METHOD VALIDATION**

## Accuracy: Recovery study

To determine the accuracy of the proposed method, recovery studies were carried out by adding different amounts (80%, 100%, and 120%) of pure drug of EVEROLIMUS were taken and added to the pre-analyzed formulation of concentration  $10\mu g/ml$ . From that percentage recovery values were calculated. The results were shown in table-2.

**Table 2: Accuracy Readings** 

Sample ID	Concentrat	ion (μg/ml)	_	% Recovery of	Statistical Analysis
Sample 1D	Pure drug	Formulation	Peak Area	Pure drug	Statistical Alialysis
$S_1: 80 \%$	8	8.03	385224	100.3	Mean = 100.9333%
S <sub>2</sub> : 80 %	8	8.08	387234	101	S.D. = 0.602771
$S_3:80\%$	8	8.12	389166	101.5	% R.S.D.= 0.59719
S <sub>4</sub> : 100 %	10	10.04	491354	100.4	Mean = 100.6666%
S <sub>5</sub> : 100 %	10	10.07	491423	100.7	S.D. = 0.251661
S <sub>6</sub> : 100 %	10	10.09	491854	100.9	% R.S.D.= 0.24999
S <sub>7</sub> : 120 %	12	12.02	526491	100.1	Mean = 100.3666%
S <sub>8</sub> : 120 %	12	12.05	542354	100.4	S.D. = 0.503322
S <sub>9</sub> : 120 %	12	12.08	561426	100.6	% R.S.D.= 0.50148

## **Precision: Repeatability**

The precision of each method was ascertained separately from the peak areas & retention times obtained by actual determination of five replicates of a fixed amount of drug. Everolimus (API) The percent relative standard deviation were calculated for Everolimus are presented in the table 3.

**Table 3: Repeatability Results** 

HPLC Injection	
<b>Replicates of Everolimus</b>	Area
Replicate – 1	162359
Replicate – 2	166447
Replicate – 3	170251
Replicate – 4	167368
Replicate – 5	170088
Replicate – 6	169387
Average	167650
Standard Deviation	3008.243
% RSD	1.79435908

## **Intermediate precision: Intra-assay & inter-assay**

The intra & inter day variation of the method was carried out & the high values of mean assay & low values of standard deviation & % RSD (% RSD < 2%) within a day & day to day variations for Everolimus revealed that the proposed method is precise.

Table 4: Results of Intra-Assay & Inter-Assay

Conc. Of Everolimus	Observed Conc. O	of Everolimus (	μg/ml) by the prop	osed method
$(API) (\mu g/ml)$	Intra-D	ay	Inter-Day	
	Mean (n=6)	% RSD	Mean (n=6)	% RSD
8	8.08	0.96	8.03	0.97
10	10.04	0.40	10.29	0.42
12	12.97	0.33	12.52	0.14

## Linearity & Range

The calibration curve showed good linearity in the range of 0-14  $\mu$ g/ml, for Everolimus (API) with correlation coefficient (R<sup>2</sup>) of 0.999. A typical calibration curve has the regression equation of y = 45801x - 17108 for Everolimus.

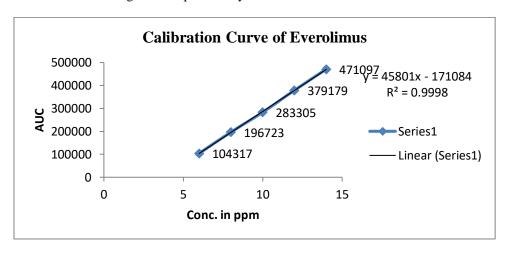


Fig 4: Calibration Curve for Everolimus

## **Method Robustness**

(Table-5.24, % RSD < 2%) the developed RP-HPLC method for the analysis of Everolimus (API).

Table 5: Result of method robustness test

Change in parameter	% RSD
Flow (1.1 ml/min)	0.02
Flow (0.9 ml/min)	0.08
Temperature (27°C)	0.04
Temperature (23 <sup>o</sup> C)	0.16
Wavelength of Detection (274 nm)	0.05
Wavelength of detection (270 nm)	0.07

## LOD & LOQ

The Minimum concentration level at which the analyte can be reliable detected (LOD) & quantified (LOQ) were found to be 0.04 &  $0.12 \,\mu\text{g/ml}$  respectively.

## System Suitability Parameter

The System Suitability parameters examined. Finally system suitability test parameters are established. The obtained data is shown in the following table 6.

Table 6: Results of System Suitability Parameter

S. No	Parameter	Everolimus
1	Retention time	2.79
2	Theoretical plates	2596
3	Tailing factor	1.07
4	Peak Area	218965
5	Resolution	3.72

#### ASSAY OF EVEROLIMUS IN DOSAGE FORM

Estimation of Everolimus in Tablet Dosage Form

Table 7: The assay of Afinitor Tablet tablets containing was found to be 99.4 %.

Brand name tablets	of Labelled amount of Drug (mg)	Mean (±SD) amount (mg) found by the proposed method (n=6)	
Afinitor Tablet	10	9.94 (±0.05)	99.4 (±0.48)

#### FORCED DEGRADATION STUDIES

The result of forced degradation studies are given in the following table-7.6.

Table 8: Results of force degradation Everolimus studies of API.

Stress condition	Time	Assay of active	Assay of degraded	Mass Balance
	(hours)	substance	products	(%)
Acid Hydrolysis (0.1N HCl)	24Hrs.	96.44	3.56	100
Basic Hydrolysi (0.IN NaOH)	24Hrs.	97.11	2.89	100
Thermal Degradation (50 °C)	24Hrs.	95.77	4.23	100
UV (254nm)	24Hrs.	96.25	3.75	100
3% Hydrogen peroxide	24Hrs.	96.79	4.21	100

#### **RESULT & DISCUSSION**

To develop a precise, linear, specific & suitable stability indicating RP-HPLC method for analysis of Everolimus, different chromatographic conditions were applied & the results observed are presented in previous chapters. Isocratic elution is simple, requires only one pump & flat baseline separation for easy and reproducible results. So, it was preferred for the current study over gradient elution. In case of RP-HPLC various columns are available, but here develosil C<sub>18</sub>, 5µm, 150 x 4.6 mm i.d. column was preferred because using this column peak shape, resolution and absorbance were good. Mobile phase & diluent for preparation of various samples were finalized after studying the solubility of API in different solvents of our disposal (methanol, acetonitrile, dichloromethane, water, 0.1N NaOH, 0.1NHCl). The drug was found to be highly soluble in DMSO, Ethanol and Dimethyl Formamide (DMF), sparingly soluble in Aqueous Buffers . Drug was insoluble in Acetonitrile and water. Using these solvents with appropriate composition newer methods can be developed and validated. Detection wavelength was selected after scanning the standard solution of drug over 200 to 400nm. From the U.V spectrum of Everolimus it is evident that most of the HPLC work can be accomplished in the wavelength range of 272 nm conveniently. Further, a flow rate of 1 ml/min & an injection volume of  $10~\mu l$  were found to be the best analysis. The result shows the developed method is yet another suitable method for assay, stability and purity which can help in the analysis of Everolimus in different formulations.

## **CONCLUSION**

A sensitive & selective RP-HPLC method has been developed & validated for the analysis of Everolimus API. Further the proposed RP-HPLC method has excellent sensitivity, precision and reproducibility.

#### REFERENCES

- 1. Patil KR, Rane VP, Sangshetti JN, Shinde DB. A Stability-Indicating LC Method for the Simultaneous Determination of Telmisartan and Ramipril in Dosage Form. Chroma. 2008;67(7-8):575-82. doi: 10.1365/s10337-008-0550-5.
- 2. Rasayan J Chem. Coden: RJCABP;5 | No.1 | 90-105 | January-March | 2012ISSN:0974-1496.
- 3. Notari S, Tommasi C, Nicastri E, Bellagamba R, Tempestilli M, Pucillo LP et al. IUBMB Life. Simultaneous determination of maraviroc and raltegravir in human plasma by HPLC-UV. 2009 Apr;61(4):470-5. doi: 10.1002/iub.181, PMID 19319971.
- 4. Baht and Leena. J Liq Chromatogr. 2007;30:309.
- 5. Williard HH, Merit LL, Dean FA, Settle FA. Instrumental methods of analysis. 7th ed, C.B.S.Publishers. New Delhi; 2002.
- 6. Menon GN, White LB, Department of Analytical Research, Abbott Laboratories, (pubmed-index for MEDLINE).
- 7. International Conference on Harmonization. Q2A: text on validation of analytical procedures. Fed Regist. Vol. 60(40); 1995. p. 11260-2.
- 8. International Conference on Harmonization. Q2B: validation of analytical procedures: methodology; availability. Fed Regist. Vol. 62(96); 1997. p. 27463-7.
- 9. FDA. Analytical procedures and methods validation: chemistry, manufacturing and controls documentation; availability. Fed Regist (Notices) 65. 2000;169:52776-7.
- 10. Shabir GA. Validation of HPLC chromatography methods for pharmaceutical analysis. Understanding the differences and similarities between validation requirements of FDA. J Chromatogr A. 2003;987(1-2):57-66. doi: 10.1016/s0021-9673(02)01536-4, PMID 12613797.

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- 11. Torrealday N. J of Pharmaceutical and biomed. Ana. 2003;32:847.
- 12. Validation of analytical procedures, Methodology. ICH harmonized tripartite guideline. Vol. 108; 1996.
- 13. FDA drug approvals list [online][cited Aug 26 2003].
- 14. Green JM. A practical guide to analytical method validation, Anal. Chem News Features. May 1 1996:305A-9A.
- 15. Winslow PA, Meyer RF. Defining a master plan for the validation of analytical methods. J Validation Technol. 1997:361-7.
- 16. Pilote L, Abrahamowicz M, Eisenberg M, Humphries K, Behlouli H, Tu JV. Effect of different angiotensin-converting-enzyme inhibitors on mortality among elderly patients with congestive heart failure. CMAJ. May 2008;178(10):1303-11. doi: 10.1503/cmaj.060068, PMCID 2335176. PMID 18458262.
- 17. Remuzzi G, Macia M, Ruggenenti P. Prevention and treatment of diabetic renal disease in Type 2 diabetes: the BENEDICT study. J Am Soc Nephrol. Apr 2006;17(4);Suppl 2:S90-7. doi: 10.1681/ASN.2005121324, PMID 16565256.
- 18. Kuhn B, Jacobsen W, Christians U, Benet LZ, Kollman PA. Metabolism of sirolimus and its derivative everolimus by cytochrome P450 3A4: insights from docking, molecular dynamics, and quantum chemical calculations. J Med Chem. 2001 Jun 7;44(12):2027-34. doi: 10.1021/jm010079y, PMID 11384247.
- 19. Krueger DA, Care MM, Holland K, Agricola K, Tudor C, Mangeshkar P, Wilson KA, Byars A, Sahmoud T, Franz DN: Everolimus for subependymal giant-cell astrocytomas in tuberous sclerosis. N Engl J Med. 2010 Nov 4;363(19):1801-11. doi: 10.1056/NEJMoa1001671.
- 20. den Burger JC, Wilhelm AJ, Chahbouni A, Vos RM, Sinjewel A, Swart EL. Analysis of cyclosporin A, tacrolimus, sirolimus, and everolimus in dried blood spot samples using liquid chromatography tandem mass spectrometry. Anal Bioanal Chem. 2012 Oct;404(6-7):1803-11. doi: 10.1007/s00216-012-6317-8. PMID 22899246.
- 21. Pawaskar DK, Straubinger RM, Fetterly GJ, Hylander BH, Repasky EA, Ma WW et al. Synergistic interactions between sorafenib and everolimus in pancreatic cancer xenografts in mice. Cancer Chemother Pharmacol. 2013 May;71(5):1231-40. doi: 10.1007/s00280-013-2117-x. PMID 23455452.
- 22. Kapavarapu S, Chintala R. Stability indicating RP-HPLC method for the estimation of everolimus in pharmaceutical formulations. Am J PharmTech Research;5(2), ISSN: 2249-3387, Pg no: 332-344.
- 23. Sharmila D. Lakshmana Rao\* and l. Kalyani, Development and validation of stability-indicating High Performance Liquid chromatographic Method for the estimation of everolimus in Tablets. Indian J Pharm Sci;77(5), Pg no: 599-604.